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Structure-based virtual screening of plant-derived natural compounds as potential PPARα agonists for the treatment of dyslipidemia

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Abstract: Background: Nowadays, metabolic disorders such as dyslipidemia have become serious health problems in the modern world. PPARs are regulators of numberous metabolic pathways, hence there has been a huge increase in the development and use of the PPARs agonists, especially PPARa agonists as main therapeutic of dyslipidemia. Objectives: The study aimed to explore potential plant-derived natural compounds as PPARa agonist agent for drug discovery of dyslipidemia. Methods: Structure-based virtual screening through molecular docking was conducted for 142 bioactive compounds from 29 medicinal plants on the main binding site of PPARa (PDB ID: 5HYK). Binding affinities and binding interactions between the ligands and PPARa were investigated. **Results:** Screening results showed that 34 compounds had strong binding affinities into the PPARα (binding affinities of less than -8.0 kcal.mol⁻¹), including 20 flavonoid, 4 terpenoid and 10 alkaloid compounds. Flavonoid was found as the best group which fitted well in the binding site of the PPARa. Top compounds were identified, including formononetin from Thermopsis alterniflora (-10.2 kcal.mol⁻¹), diosmetin from *Musa spp.* (-10.1 kcal.mol⁻¹), luteolin from *Elsholtzia ciliate* (-9.9 kcal.mol⁻¹); steviol from Stevia rebaudiana (-9.4 kcal.mol⁻¹); and tuberocrooline from Stemona tuberosa (-10.5 kcal.mol⁻¹), respectively. These compounds showed the potential agonistic activities due to forming the hydrogen bonds as well as hydrophobic interactions with four key residues of the receptor such as Ser280, Tyr314, His440 and Tyr464. Conclusions: These potential natural compounds may provide useful information in the drug design and discovery for anti-dyslipidemia agents.

Keywords: structure-based virtual screening; molecular docking; PPARa; natural compounds; dyslipidemia.

1. INTRODUCTION

Nowadays, with the change of living standards and lifestyle, metabolic disorders such as dyslipidemia (or often hyperlipidemia) and obesity have become serious health problems in the modern world [1]. These diseases can lead to high incidence of morbidity and mortality in both men and women as well as cause an economic burden to the society [1]. Dyslipidemia is characterized by a high levels of total or lowdensity lipoprotein (LDL) cholesterol, elevated triglycerides

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and/ or a low levels of high-density lipoprotein (HDL) cholesterol [2]. Dyslipidemia is considered as the biggest contributing factor to the development of athrosclerosis and cardiovascular diseases which is the first cause of death in both developed and developing countries [2]. According to the World Health Organisation, there are about 50% of patients of ischemic heart disease associated with dyslipidemia and more than 4 million deaths every year [3].



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Peroxisome Proliferator-Activated Receptors (PPARs) are the nuclear hormone receptors superfamily (class II) [4] including three subtypes, PPAR α , PPAR β or δ , and PPAR γ , respectively [5]. The PPARs are ligand-activated transcription factors which play a crucial role in the regulation of metabolic processes, energy homeostatis by inducing or repressing target genes [6]. Each subtype displays distinct functions in different cell types [5]. The PPARa was known for controlling and regulating lipid metabolism and inflammation [7]. This receptor is expressed in tissues which have high rate of fatty acid catabolism and in tissues involved in lipid oxidation such as cardiac muscle, liver, kidney, skeletal and adrenal glands [8]. The PPAR δ appears in skeletal muscle and adipose tissues which is best known for skin homeostatis [8] and for regulatation of cholesterol, adipogenesis metabolism, and colon cancer [9]. The PPARy is present in adipose tissue, vascular smooth muscles and immune cells repsonible for energy storage [5] by inducing lipogenesis and fat storage in the tissue as well as improving insulin sensitivity in skeletal muscle [10].

All three PPARs are regulators of numberous metabolic pathways, hence there is a huge increase in the development and use of the PPARs agonists, especially PPARa agonists as main therapeutics of dyslipidemia during the last decade [8]. Number of current marketed drugs for dyslipidemia treatment targets the PPARs. For example, fibrates drugs are activators of PPARa to reduce triacylglycerols used for the treatment of dyslidemia and thiazolidinediones as activatiors of PPARy for treating hyperglycemia in the type-2 diabetes. Beside, there are other molecular targets for anti-dyslipidemia such as inhibitions of cholesteryl ester transfer protein, cholesterol absorption by binding to cholesterol transporter NPC1L1 (Niemann-pick C1-like1) protein, cholesterol Oacyltransferase enzymes involved in re-esterification of absorbed cholesterol, cholesterol-metabolizing cytochrome P450 or activtion of AMP-activated protein kinase and omega-3 fatty acids, etc [8]. However, there is still rarely drug compounds for those targets [8]. Thus, the PPARs, in particular PPARa has been the main target for antidyslipidemia.

The three-dimensional structures of human PPAR α in complex with several agonists were solved (PDB codes: 5HYK, 4BCR, 3SP6, 3KDT, 3G8I, 3FEI, 3ET1, 2REW, 2GTK, 117G). The structure of human PPAR α is very similar to both of the PPAR γ and PPAR δ [5]. The structure includes four functional domains (A/B, C, D, E/F) (5), in which there are the ligand-independent activation function 1 (AF1) in the A/B domain; the conserved central DNA binding domain in the C domain; docking site for cofactors in the D domain and the ligand binding domain in the E region [5]. The PPARs have multiple binding sites depending on various binding ligands [11]. There are two known binding sites, the main binding pocket for full agonistic and the second one for partial agonistic activities on the PPAR α [12].

The ligand binding domain of PPAR α is a 12-helix which forms a large hydrophobic pocket [5] (T-shaped cavity) with the volume of about 1300 Å³ [13]. The central pocket spans the region located in between the C-terminal helix 12 (forming one side of a second activation function, AF2 helix) and the 3-stranded antiparallel β sheet [13]. After ligation with the agonist such as fibrate drugs, PPAR α undergoes conformation changes [6, 14]. The conserved hydrogen bonds involving the AF2 helix were formed between the protein and the agonists which was found as the most structural differences between the apo protein and the agonist-bound protein [13]. The full agonists stabilise an active conformation of the AF2 helix and enhances heterodimerization process with the retinoid X receptors, promoting recruitment of nuclear receptor coactivators and gene transcription [12, 13]. In addition, a loss of this stabilizating interactions suggested the partial agonist [14]. Without ligands, the PPAR α binds to promoters of the target genes leading to transcription repress [6].

However, fibrate drugs as PPARa agonists demonstrated to cause some side effects such as hepatomegaly or liver dysfunction, gastrointestinal disturbance, increase of creatinine levels, etc. [15]. As the results, there is necessary to explore novel compounds for the anti-hyperlipidemic activity with less toxicities. Some medicinal plants with bioactive compounds have been reported for anti-lipid effects as rich source for new effective and safe medicines [16-20] despite of required sufficient evidence for their activities. For example, anthraquinones from Rheum officinale; ginsenoside, ginseng, and polysaccharides from Radix ginseng; triterpenes from Rhizoma alismatis,... reduced triglycerides, LDL-C [19]. Furthermore, some compounds were reported as PPAR α agonists such as picrasidine C (an alkaloid from the root of Picrasma quassioides); bixin (a carotenoid from the pericarp of the seeds of Bixa orellana); naringenin (flavanones from the dried, immature fruit of Citrus aurantium); secoiridoid excelside B and some metabolites from Fraxinus excelsior L.etc [20]. Therefore, this study aimed to discover the plantderived natural compounds against PPARa as PPARa agonists to assist in drug design for anti-dyslipidemia. By using molecular docking for screening, 142 investigated compounds from 29 medicinal plants were docked into the PPAR α . The results of ligand binding affinities and their interactions with this receptor were combined to select the potential natural compounds as potential lipid lowering agents.

2. MATERIALS AND METHOD

Molecular docking was applied in *in silico* screening to select natural compound with high binding affinity into the PPAR α . Initially, the diverse bioactive compounds belonging to four main groups, flavonoid, terpenoid, alkaloid and saponin, respectively with their lower-lipid effects were selected for this study. To assist for drug discovery in the next step, the medicinal plants were then searched for containing the compounds. In total, there were 142 natural compounds, flavonoid (36 compounds), terpenoid (33 compounds), flavonoid (36 compounds) were chosen [20-27]. Molecular docking was conducted to investigate binding interactions and to reveal potential natural compounds for treatment of lipid disorders using AutoDock Vina software version 1.1.2 (an open-source program) [28].

Preparation of protein: The 3D crystal structure of the complex PPAR α was retrieved from the Protein Data Bank (PDB ID: 5HYK – resolution: 1.83 Å, https://www.rcsb.org) [6]. The structure of PPAR α has a co-crystallized ligand-2-methyl-2-[4-(naphthalen-1-yl)phenoxy]propanoic acid (or AL26-29) as the full agonist [6]. The residues making up the main binding pocket are Cys276, Thr279, Ser280, Tyr314, Ile317, Met330, His440 and Tyr464 [6] which covered the co-crystallised-ligand. Using AutoDock Tools 1.5.7rc1, this

receptor was prepared for docking; with removing all water molecules and heteroatoms and adding polar hydrogen atoms and Kollman charges into the protein structure. Redocking was carried out by extracting the co-crystallized ligand from the experimental structure and docking the ligand into the binding pocket of PPAR α encompassing the native ligand.

Preparation of ligand: Ligands (142 compounds) were prepared in 2D by Chem3D Ultra program (CDX file). Open Babel was employed to convert all 2D structures to 3D structures whose energies were subsequently minimized with the YASARA Energy Minimization server (https://www.yasara.org).

Docking parameters: The docking parameters included the coordinate parameters of the center x, y, z of 7.57 Å; 32.34 Å; 23.882 Å. The grid box was centered on the native ligand with the dimension of $24 \times 24 \times 24$ (Å)³; spacing distance = 1 Å; with default exhaustiveness = 8.

Evaluation of docking results: The docking results were evaluated by the ligand binding affinities (kcal.mol⁻¹), binding pose and the possible interactions between the key residues for biological activities in the target such as Tyr314 for maintaining protein active conformation of protein and for selectivity ligands. In addition, the fibrate drugs as the PPAR α agonists including fenofibrate, gemfibrozil, clofibrate, bezafibrate, ciprofibrate were used for docking into the active site of the receptor as reference drugs. The lower binding affinity, the better ligand. BIOVIA Discovery Studio Visualizer 2020 (a free version, downloaded from the website https://discover.3ds.com/) was used for visualisation, assisting in analysis and creating images of 3D models of the protein and the binding mode of protein-ligand.

3. RESULTS

To evaluate the docking protocol, redocking of the cocrystallized ligand into the main binding site of PPARa was conducted. The results showed that this native ligand (the naphthalenic derivatives) bound well into the PPAR α with good binding affinity (-11.3 kcal.mol⁻¹) and the root mean square deviation (RMSD) between the docked structure and the native one, using only movable heavy atoms (i.e., only ligand atoms, not hydrogen), was 1.46 Å (less than 2.00 Å). The binding mode and interactions of this ligand with the PPAR α were mostly similar to the experimental structure. For example, the polar head of the ligand could form the hydrogen bonds with the side chains of key residues of the active site of the PPARa such as Tyr464 on the H12 helix responsible for agonistic activities towards PPARa and Tyr314, and with the OH of Ser280 [6]. The ligand also interacted with the PPARa through hydrophobic interactions with Phe273 and Phe351. These results demonstrated the reliability of the docking program, so the docking protocol could be then used for screening process.

Table 1. Screening results of 142 natural compounds based on molecular docking from 29 medicinal plants into the PPAR α (PDB id: 5HYK) with the binding affinities (kcal.mol⁻¹).

| No | Medicinal plant | Natural compound | Binding affinity (kcal.mol ⁻¹) | No | Medicinal plant | Natural compound | Binding affinity (kcal.mol ⁻¹) |
|----|---|----------------------|--|----|---|--|--|
| 1 | <i>Momordica charantia</i> Cucurbitaceae | Momordicine 28 | -1.8 | | Cynachium | Hancolupenone | -1.4 |
| | | Gypsogenin | -3.7 | 7 | stauntonii Apocynaceae | Hancolupenol | -2.1 |
| | | Goyaglicoside A | -6.9 | 8 | Folium nelumbinis Nuciferae | Nuciferin | -5.3 |
| | | Goyaglicoside C | -7.2 | | | Pronuciferine | -4.0 |
| | | Goyaglicoside E | -8.2 | | | Roemerine | -7.4 |
| | | Goyaglicoside F | -7.2 | | | 2-hydroxy-1-methoxyaporphine | -6.9 |
| | | Vicine | -6.9 | | | Dehydronuciferine | -2.6 |
| | | Goyaglicoside H | -6.0 | | | N-nornuciferine | -3.5 |
| 2 | <i>Stevia rebaudiana</i> Asteraceae | Steviol | -9.4 | - | | Astragalin | -3.8 |
| | <i>Ganoderma lucidum</i> Ganodermataceae | Ganoderic acid A | -1.2 | 9 | <i>Erythrina orientalis</i> Fabaceae | Soyasapogenol B | -3.2 |
| | | Ganoderic acid B | -1.1 | | <i>Codonopsis pilosula</i> Campanulaceae | Radicamine A | -8.2 |
| | | Ganoderic acid D | -1.5 | 1 | | Codonopyrrolidiums A | -5.9 |
| | | Ganoderic acid E | -3.8 | 10 | | Codonopyrrolidiums B | -7.1 |
| | | Ganoderic acid F | 4.0 | | | Codonopsinols A | -7.9 |
| | | Ganoderic acid G | -0.3 | | | Codonopsinols B | -7.6 |
| 2 | | Ganoderic acid H | -5.3 | | | Codonopsinols C | -7.4 |
| 3 | | Ganoderic acid I | -3.9 | | | Codonopiloside A | -7.5 |
| | | Ganoderic acid J | -2.7 | | | Tryptophan | -7.7 |
| | | Ganoderic acid K | 5.9 | | | Perlolyrine | -9.1 |
| | | Ganoderic acid L | -0.8 | | | Nicotinic acid | -5.5 |
| | | Ganoderic acid M | -0.2 | | <i>Stemona tuberosa</i> Stemonaceae | Isostenine (neostenine) | -5.4 |
| | | Ganoderic acid N | -5.0 | | | Tuberostemonine H | -2.3 |
| | | Acid ganodermic S | -0.5 | | | Tuberostemonine N | -2.6 |
| | | Acid ganodermic P2 | -0.2 | | | Tuberostemonine K | -6.1 |
| | Cynara scolymus Asteraceae | Cynaropicrin | -8.9 | | | Neotuberostemonol | -4.2 |
| 4 | | Dehydrocynaropicrin | -8.3 | 11 | | Epi-Bisdehydroneotuberostemonine J (aka epibisdehydrotuberostemonine J) | -8.5 |
| | | Grossheimin | -7.0 | 11 | | 9a-Bisdehydrotuberostemonine | -1.0 |
| 5 | Sesamum indicum Pedaliaceae | 3-epibartogenic acid | -2.3 | | | 9a-Bisdehydrotuberostemonine A | -7.0 |
| | | Celasdin A | -2.1 |] | | Tridehydrotuberostemonine | -2.0 |
| 6 | <i>Celastrus hindsii</i> Celastraceae | Celasdin B | -0.9 |] | | Bisdehydroneostemoninine | -9.1 |
| | | Celasdin C | -6.0 |] | | Bisdehydrostemoninine A | -8.5 |
| | | Maytenfolone A | -0.6 | | | Bisdehydrostemoninine B | -8.5 |

Table 1. (continue)

| No | Medicinal plant | Natural compound | Binding affinity (kcal.mol ⁻¹) | No | Medicinal plant | Natural compound | Binding affinity (kcal.mol ⁻¹) |
|----|--|--|--|----|---|-------------------------|--|
| | | Bisdehydrostemoninine | -8.1 | 19 | Lysimachia foenum- graecum Primulaceae | Foenumoside B | -1.3 |
| | | Isobisdehydrostemoninine | -7.3 | 20 | Dioscorea nipponica Makino Dioscoreaceae | Trillin | -2.9 |
| | | Oxystemoninine | -3.7 | 21 | Panax ginseng Araliaceae | Ginsenoside Rb1 | -5.5 |
| | | Stemoenonine | -5.0 | | <i>Scutellaria baicalensis</i> Lamiaceae | Formononetin | -10.2 |
| | | 9a-O-Methylstemoenonine | -4.3 | 22 | | Daidzein | -9.7 |
| | | Oxystemoenonine | -4.0 | | | Chrysin | -9.6 |
| | | Tuberostemospironine | -7.5 | | | Isoscitellarein | -9.4 |
| | | 10-Hydroxycroomine | -7.5 | | | Daidzin | -9.1 |
| 11 | Stemona tuberosa | 6-Hydroxycroomine (aka 6ahydroxycroomine) | -7.6 | | | Apigenin-7-glucoside | -8.9 |
| | Stemonaceae | Dehydrocroomine | -5.8 | | | Luteolin-7-rutinoside | -7.2 |
| | | Tuberospironine | -7.6 | | | Puerarin | -5.9 |
| | | Neotuberostemoninol | -6.7 | | | Luteolin | -9.9 |
| | | Sessilifoliamide F | -2.5 | 22 | Elsholtzia ciliate | Apigenin | -9.4 |
| | | Tuberostemoline | -0.7 | 23 | Lamiaceae | Kumatakenin | -8.1 |
| | | Tuberocrooline | -10.5 | | | Linarin | -6.5 |
| | | 1,9a-seco-Stemoenonine | -6.2 | | | Epicatechin | -8.9 |
| | | Tuberostemoenone | 0.9 | 24 | <i>Musa spp.</i> Musaceae | Gallocatechin | -8.9 |
| | | Croomine | -8.9 | | | Dimer procyanidin | -7.7 |
| | | Stemoninoamide | -9.8 | | | Semilicoisoflavone | -7.7 |
| | | Stemotinine | -6.1 | | | Chalcone | -8.9 |
| | | Tuberostemonone | 2.9 | 25 | | Isoliquiritigenin | -8.6 |
| | | Neotuberostemonine (aka tuberostemonine LG) | -1.0 | | <i>Glycyrrhiza</i> uralensis Fabaceae | Ononin | -8.6 |
| 10 | Platycodon grandiflorum Campanulaceae | Platycodin A | -4.5 | | | Licoflavonol | -8.4 |
| | | Platycodin C | -4.8 | | | Isoangustone | -8.2 |
| 12 | | Platycodin D | -3.9 | | | Licochalcone | -7.7 |
| | | Deapioplatycodin D | -4.1 | | | Glyasperin C | -7.6 |
| 13 | Kochia scoparia Chenopodiaceae | Momoridin Ic | -0.9 | | | 7-O-methylluteone | -7.1 |
| | Aesculus turbinate Aesculaceae | Escin Ia | -2.5 | _ | | Licorisoflavan A | -6.9 |
| | | Escin IIa | -2.2 | | | Dehydroglyasperin | -6.3 |
| 14 | | Escin Ib | -3.4 | 26 | <i>Cynanchum</i> Apocynaceae | Quercetin | -9.2 |
| | | Escin IIb | -3.9 | 27 | Thermopsoside alternilora Fabaceae | Thermopsoside | -8.8 |
| | Fructus Momordicae grosvenorii Cucurbitaceae | Mogroside IV | -1.2 | | | Saponarin 4-O-glucoside | -8.7 |
| 15 | | Mogroside V | -0.7 | 28 | <i>Cucumis sativus</i> Cucurbitaceae | Isovitexin | -7.2 |
| | Schefflera heptaphylla Araliaceae | Silphioside F | -2.1 | | | Isoorientin | -68 |
| 16 | | Conteroside B | -37 | | | Saponarin | 15 |
| | | Gypsogenin 3- <i>O</i> -D- glucuronide | -0.5 | | | Vicenin-2 | 2.5 |
| 17 | Schefflera octophylla Araliaceae | Sessiloside | -6.2 | 29 | <i>Senna alata</i> Fabaceae | Diosmetin | -10.1 |
| | | Chiisanoside | -7.1 | 1 | | | |
| | Gynostemma | Rutin | -3.4 | | | | |
| 18 | pentaphyllum | Damulin A | -7.4 |] | | | |
| | Cucurbitaceae | Damulin B | -6.7 | 1 | | | |

Molecular docking

All of 142 natural compounds of the medicinal plants were located into the main binding pocket of PPAR α (PDB id: 5HYK) in docking which showed their potential activities against PPAR α . These compounds were classified into the subgroups: alkaloid (51 compounds), terpenoid (33 compounds), flavonoid (36 compounds) and saponin (22 compounds) for binding analysis. The results were also compared with those results obtained with the agonisit ligands such as fibrate drugs as reference compounds of PPAR α . It showed that flavonoid was the best group of binding well into the PPAR α compared to the other groups. Notably, total of 34/142 natural compounds had high affinity for binding into the PPAR α (binding affinities of less than -8.0 kcal.mol⁻¹), including 20 flavonoid compounds, 4 terpenoid compounds and 10 alkaloid compounds (Table 1). Of which, the top compounds for each group included: flavonoid group: formononetin from *Thermopsis alterniflora* (-10.2 kcal.mol⁻¹), diosmetin from *Musa spp*. (-10.1 kcal.mol⁻¹), luteolin from *Elsholtzia ciliate* (-9.9 kcal.mol⁻¹); one terpenoid compound: steviol from *Stevia rebaudiana* (-9.4 kcal.mol⁻¹); and one alkaloid compound: tuberocrooline from *Stemona tuberosa* (-10.5 kcal.mol⁻¹) (Figure 1 and 2). These compounds strongly accommodated and formed good interactions with the residues of the PPAR α binding site as the reference drugs did. Structure and binding affinity relationships of the ligands were also taken for analysis. However, there were four compounds, namely saponarin, vincenin-2, stemonone and tuberostemoenone did not show negative binding affinities as the ligands did not accommodate fully in the binding pocket.



Figure 1. Binding modes of 3 top flavonoid compounds, namely formononetin from *Thermopsis alterniflora* (-10.2 kcal.mol⁻¹), diosmetin from *Musa spp*. (-10.1 kcal.mol⁻¹), luteolin from *Elsholtzia ciliate* (-9.9 kcal.mol⁻¹); respectively into the PPARα (PDB id: 5HYK) with green line represented for hydrogen bond, pink line for hydrophobic contacts



Figure 2. Binding modes of 2 top compounds, one terpenoid compound: steviol from *Stevia rebaudiana* (-9.4 kcal.mol⁻¹); and one alkaloid compound: tuberocrooline from *Stemona tuberosa* (-10.5 kcal.mol⁻¹), respectively into the PPARα (PDB id: 5HYK) with green line represented for hydrogen bond, pink line for hydrophobic contacts

Binding modes and interactions analysis

Flavonoid compounds such as formononetin (-10.2 kcal.mol-¹) and diosmetin (-10.1 kcal.mol⁻¹) had good binding affinities to the binding pocket of PPARa which showed the potential agonistic activities. This was explained by the presence of -OH groups in the structure of these flavonoids created the hydrogen bonds with four key residues of the receptor such as Ser280. Tyr314, His440 and Tyr464 [6] as well as the aromatic rings made hydrophobic interactions with the target residues which led to the tightly attachment of the compounds into the binding site of PPARa. These results were compatible with the experimental results that PPAR agonists interact to form similar interactions with polar residues, especially with Tyr464 localized in AF2 which was important for stabilizing the active conformation of protein [6]. In addition, the three other flavonoids such as luteolin (-9.9 kcal.mol⁻¹), daidzein (-9.7 kcal.mol⁻¹), and chrysin (-9.6 kcal.mol⁻¹) were also good ones towards the binding affinities on the PPAR α . The ligands with the simple backbone could attach the hydrophobic pocket better than the isoflavon and flavon compounds with the presence of glucosides or alkyl groups, for example: luteolin (-9.9 kcal.mol⁻¹) > luteolin-7-rutinoside (-7.2 kcal.mol⁻¹); daidzein (-9.7 kcal.mol⁻¹) > daidzin (-9.1 kcal.mol⁻¹); and apigenin (-9.4 kcal.mol⁻¹) > apigenin-7-glucoside (-8.9 kcal.mol⁻¹). The results could be related to the experimental resuls that the aglycone penetrates easily due to the high lipophilicity and low molecular weight [29]. The other isoflavon and flavon compounds with the presence of glucoside or alkyl groups such as daidzin, thermopsoside, ononin, isoangustone, apigenin-7glucoside, saponarin 4-O-glucoside, licoflavonol, kumatakenin did not show strong binding affinities (Table 1). Structures of the eight compounds with the good binding affinities of less than -8.0 kcal.mol⁻¹ and their binding modes into the PPARα (PDB id: 5HYK) were presented in Figure 3 and 4.



Figure 3. Structures of 8 flavonoid compounds belonging to isoflavon and flavon with the good binding affinities of less than -8.0 kcal.mol⁻¹ into the PPARα (PDB id: 5HYK)



Figure 4. Binding modes of 8 flavonoid compounds belonging to isoflavon and falvon with the good binding affinities of less than -8.0 kcal.mol⁻¹ into the PPAR α (PDB id: 5HYK)

Alkaloid compounds bound to the PPARa with binding affinities ranging from -1.0 to -10.5 kcal.mol⁻¹. Ten alkaloid compounds including tuberocrooline (-10.5 kcal.mol⁻¹ had good affinities on the PPARa due to forming the hydrophobic interactions between the aromactic rings containing nitrogen and some important amino acids of binding site such as His440 (for example, periolyrine: -9.1 kcal.mol⁻¹). The compounds also generated the hydrogen bonds through their side chains having some functional groups (OH or COOH groups) with residues Ser280, Tyr314 and His440 of PPARa (for example, radicamine A: -8.2 kcal.mol⁻¹). These interactions were shared by all PPARα agonist like fibrates and the other ligands reported in the PDB [6, 12], with data showed in Table 2. The alkaloid possess the simple aromactic ring like radicamine A and perlolysine could attach further in the binding pocket than the ligands having bulky aromatic groups. However, replacing pyrrole ring into pyrrolidine ring made the compound not going deeper into the cavity or extending the length of side chain of pyrrole which led to reduce the binding affinities, such as the case of compound 9abisdehydrotuberostemonie with the pyrrole ring (-7.0 kcal.mol⁻ 1) > neotuberostemonol with the pyrrolidine ring (-4.2 kcal.mol⁻¹) ¹), and stemoninoamide (-9.8 kcal.mol⁻¹) > bisdehydrostemonine A and B (-8.5 kcal.mol⁻¹). When pyrrolidine opens, for example tuberocrooline (-10.5 kcal.mol⁻¹), the compound made the C=O group of tetrahydrofuran interacted with Tyr314 and also formed more hydrogen bond with the residue Ala455 of target, so this compound showed strongly affinity than, croomine (-8.9 kcal.mol⁻¹). All of Figure 5, 6 and 7 illustrated 10 top alkaloid compounds. namelv radicamine A. perlolyrine, epibisdehydroneotuberostemonine J, bisdehydroneostemoninine, stemoninoamide, bisdehydrostemoninine Α, bisdehydrostemoninine B. bisdehydrostemoninine, tuberocrooline, croomine, respectively with their good binding affinities of less than -8.0 kcal.mol⁻¹ and their binding modes into the PPARa.

Table 2. Binding interactions of 15 compounds including 10 top alkaloid compounds, namely radicamine A, perlolyrine, epibisdehydroneotuberostemonine J, bisdehydroneostemoninine, stemoninoamide, bisdehydrostemoninine A, bisdehydrostemoninine B, bisdehydrostemoninine, tuberocrooline, croomine, and 5 reference compounds (fenofibrate, gemfibrozil, clofibrate, bezafibrate and ciprofibrate), respectively and the PPARα (PDB id: 5HYK)

| No | Compound name | Binding affinity | Hydrogen honds | Hydrophobic |
|-----|-------------------------------------|---------------------------|--|-------------------------|
| 110 | Compound nume | (kcal.mol ⁻¹) | nyurogen vonus | interactions |
| 1 | Radicamine A | -8.2 | Ser280, His440, Tyr464 | Ile354, Ile447, Ala454 |
| 2 | Doulolymine | 0.1 | Dha272 Cur276 | Ile354, His440, Val444, |
| Z | renorynne | -9.1 | File273, Cys270, | Ile447, Ala454 |
| 2 | Eni Diadahudranaatuharaatamanina I | 0 5 | $S_{2}=280$ True 214 | Ile354, Val444, Ile447, |
| 3 | Epi-Bisdenydroneotuberostenionine J | -0.3 | Sei280, Ty1514 | Lys448, Leu456 |
| 4 | Bisdehydroneostemoninine | -9.1 | | Ala455 |
| - | | o r | G B C C C C C C C C C C | Ile354, Val444, Ile447, |
| 5 | Bisdehydrostemoninine A | -8.5 | Ser280, H1s440 | Leu456 |
| | | | ~ | Ile354, Val444, Ile447, |
| 6 | Bisdehydrostemoninine B | -8.5 | Cys276 | Leu456 |
| _ | | 0.1 | ···· · · · · · · · · · · · · · · · · · | Phe273. Ile354, Val444. |
| 7 | Bisdehydrostemoninine | -8.1 | H1s440, Leu456 | Ile447, Leu456 |
| 8 | Tuberocrooline | -10.5 | Ser280, Tyr314, Ala455 | , |
| 9 | Croomine | -8.9 | Ser280, His440 | Ile354, Val444, Ile447 |
| 10 | Stamoningamida | 0.9 | | Ile354, His440, Val444, |
| 10 | Stemoninoamide | -9.8 | | Ile447, Leu456 |
| 11 | Fonofibrata | 10.2 | Sar 280 Hist 40 | Phe273, Ile354, Val444, |
| 11 | renonitiate | -10.2 | Sei280, His440 | Ile447, Ala454, Leu456 |
| | | | | Val270, Phe351, |
| 12 | Comfibrozil | 82 | His/1/0 | Val444, Ile447, Lys448, |
| 12 | Genniolozh | -0.2 | 1115440 | Ala454, Ala455, |
| | | | | Leu456 |
| 12 | Clofibrata | 73 | | Phe273, Cys276, |
| 15 | Ciolibrate | -7.5 | | Ile354, Met355 |
| 14 | Bezofibroto | 0.1 | Ser280, Tyr314, His440, | 110354 Vol444 Alo454 |
| 14 | Dezanorate | -9.1 | Val444 | nc554, vai444, Ala454 |
| | | | | Val270, Phe273, |
| 15 | Ciprofibrate | -8.6 | Ser280, Tyr314, His440 | Phe351, Val444, Ile447, |
| | | | | Ala454 |



Figure 5. The molecular structures of 10 top alkaloid compounds, namely radicamine A, perlolyrine, epibisdehydroneotuberostemonine J, bisdehydroneostemoninine, stemoninoamide, bisdehydrostemoninine A, bisdehydrostemoninine, tuberocrooline, croomine, respectively with their binding affinities into the PPARα (PDB id: 5HYK)



Figure 6. Binding modes of 10 top alkaloid compounds, namely radicamine A, perlolyrine, epibisdehydroneotuberostemonine J, bisdehydroneostemoninine, stemoninoamide, bisdehydrostemoninine A, bisdehydrostemoninine B, bisdehydrostemoninine, tuberocrooline, croomine, respectively with the good binding affinities of less than -8 kcal.mol⁻¹ into the PPARα (PDB id: 5HYK)



Figure 7. Binding interactions of 10 top alkaloid compounds, namely radicamine A, perlolyrine, epibisdehydroneotuberostemonine J, bisdehydroneostemoninine, stemoninoamide, bisdehydrostemoninine A, bisdehydrostemoninine B, bisdehydrostemoninine, tuberocrooline, croomine, respectively and the residues at the binding site of the PPARα (PDB id: 5HYK) with green line represented for hydrogen bond, pink line for hydrophobic contacts

For terpenoids and alkaloids, the compounds belonging to these groups could get into the active site of the PPAR α , however, their binding affinities were not as good as the other groups. Of which, three terpenoid compounds, namely steviol, cynarpicrine and dehydrocynaropicrine and goyaglycoside E interacted well with the residues Tyr314, Tyr464, Ile447 at the active site of the PPAR α by hydrogen bonding with their affinities of -9.4; -8.9, -8.3 and -8.2 kcal.mol⁻¹, respectively (presented in Figure 8). It could be seen that the presence of OH group in the structures such as steviol, cyanopicrin, dehydrocyanopicrin increased the binding affinities. The results agreed with the experimental data that -OH group was essential for the activities in terpenoids (20). Only one saponin compound, diosgenin displayed good binding affinity of -7.5 kcal.mol⁻¹. The saponin ginsenosides, such as ginsenoside Rb1 did not show high binding affinity (-5.5 kcal.mol⁻¹). The experimental results also reported that the compound responsible for the bioactivities of ginseng which demonstrated the inhibition of PPARa in vitro and in vivo with a consequent increased serum concentrations of total cholesterol, triglycerides, and HDL cholesterol [30].

Moreover, the docking results of fibrate drugs as the PPAR α agonists (fenofibrate, gemfibrozil, clofibrate, bezafibrate, ciprofibrate) into the PPAR α as reference drugs showed that these drugs bound closely to the binding site of the receptor with high binding affinities: fenofibrate (-10.2

kcal.mol⁻¹), gemfibrozil (-8.2 kcal.mol⁻¹), clofibrate (-7.3 kcal.mol⁻¹), bezafibrate (-9.1 kcal.mol⁻¹), ciprofibrate (-8.6 kcal.mol⁻¹) (Table 2). Comparison of docking results of the 142 natural compounds and the reference drugs, it showed that most of the investigated compounds bound to the main binding pocket and strongly mimicking the observed interactions like the reference ones did. The reference drugs interacted well with the receptor by hydrogen bonds and hydrophobic interactions with some important residues such as Tyr464 important for recruiting co-activators (6) and Phe273, Tyr314, Phe351 and Tyr464. In particular, the standard polar interactions with Ser280, Tyr314 and Tyr464 presented in the all agonists complexed to the receptor (6) which were found in all of the top investigated compounds.

Overall, top compounds selected from each group including formononetin, diosmetin, luteolin, daidzein, chrysin (flavonoid compounds); steviol (terpenoid compound) and tuberrocrooline (alkaloid compound) possess high binding affinities. They occupied the same position and shared ability to form interactions with the residues of binding pocket of the PPAR α . However, taking into account of forming hydrogen bonds network with the key residues Tyr464, His440, Tyr314 and Ser280 for agonistic activities, the following compounds, namely formononetin from *Thermopsis alterniflora* (-10.2 kcal.mol⁻¹), diosmetin from *Musa spp.* (-10.1 kcal.mol⁻¹), luteolin from *Elsholtzia ciliate* (-9.9 kcal.mol⁻¹); steviol from

Stevia rebaudiana (-9.4 kcal.mol⁻¹); and tuberocrooline from *Stemona tuberosa* (-10.5 kcal.mol⁻¹) were selected. They

could be used as the potential PPAR α agonists for further biological testings for their anti-dyslipidemia effects.



Figure 8. Binding modes of 4 top terpenoid compounds, namely steviol, cynaropicrin, dehydrocynaropicrin, goyaglicoside E, respectively with the good binding affinities of less than -8.0 kcal.mol⁻¹ into the PPARα (PDB id: 5HYK)

4. DISCUSSION

The natural compounds from medicinal plants have been of interest in drug discovery for the treatment of many diseases, including dyslipidemia in recent years. Therefore, in this study, structure-based virtual screening through molecular docking was conducted for 142 investigated natural compounds into the PPARa (PDB id: 5HYK) to discover potential compounds for anti-dyslipidemia. Through the criteria including ligand binding affinities, binding modes and binding interactions, flavonoid compounds performed the most potential compounds. These compounds formed favorable bindings with the PPARa by hydrogen bonds with key residues of the target, Ser280, Tyr314 and Tyr464, crucial for agonistic activities. Of which, two flavonoids, formononetin from Thermopsis alterniflora (-10.2 kcal.mol-¹), diosmetin from *Musa spp*. (-10.1 kcal.mol⁻¹) and luteolin from *Elsholtzia ciliate* (-9.9 kcal.mol⁻¹) were the top compounds of high binding affinities which suggesting they might be potential compounds as PPAR α agonists for further testing toward anti-dyslipidemia effects.

This results were consistent with the reported results from the experiments that formononetin was the most potent activators of PPAR α with the value of EC₅₀ of less than 1.0 mol/L [20, 31]. Diosmetin demonstrated inhibition of fat accumulation in liver and epididymal tissues, however diosmetin had little effect on the parameters related to dyslipidemia [32]. Diosmetin (3',5,7trihydroxy-4'-methoxy-flavone), a flavone aglycone have been displayed to have several pharmacological effects such as antiinflammatory and hepatoprotective role in an *in vivo* model [33] or antihyperglycemic activities [34]. Another study reported that diosmetin treatment inhibited fat accumulation in liver and epididymal tissues and improved glucose intolerance by lowering glucose levels during a glucose tolerance test [32]. Furthermore, luteolin has been shown to have anti-inflammatory activity as well as to protect against tumor necrosis factor-alpha and especially to prevent lipid accumulation and decrease total

cholesterol, triglyceride and LDL-cholesterol levels [35]. Therefore, the findings in this study suggested the mechanism of these compounds toward to PPAR α .

Moreover, two other compounds, one terpenoid compound: steviol from *Stevia rebaudiana* (-9.4 kcal.mol⁻¹); and one alkaloid compound: tuberocrooline from Stemona tuberosa (-10.5 kcal.mol⁻¹) were also identified as the hit natural compounds for PPARa agonists. Steviol, the terpene aglycone, the final metabolite of all steviol glycosides was reported with some bioactivites such as antidiabetic effects on streptozotocin-induced diabetic mice [36], antihyperglycemic, antihypertensive, antioxidant and anti-inflammatory effects, etc [37]. Steviol causes a decrease in glucose accumulation in intestinal ring tissue, liver and kidney and also enhances insulin secretion [36]. Tuberocrooline has not been demonstrated for anti-dyslipidemia, but this compound was known for the antitussive activity [38]. Furthermore, the formononetin has been proved for anticancer activity [39], diosmetin for the biological activity on alloxan diabetic rats [40], steviol for antioxidant capacity [41] and tuberocrooline for the acetylcholinesterase inhibitory activity [42].

Therefore, the good compounds were selected, namely formononetin from Thermopsis alterniflora, diosmetin from Musa spp., luteolin from Elsholtzia ciliate; steviol from Stevia rebaudiana; and tuberocrooline from Stemona tuberosa, respectively for further steps in drug discovery and design of PPARα agonists for anti-hyperlipidemia activities. Combination of the experimental results of these compounds for dyslipidemia effects and the results obtained in this study, the current research also suggested the mechanism of these compounds towards to the PPARa for agonistic activities. However, there were still some limits in this study. It is required more number of investigated compounds and partial agonistic on the PPARa or dual agonist/ pan-agonistic activities combining on PPARy could be investigated in the screening process of better anti-dyslipidemia and also for PPARα agonists compounds.

Conclusion

Metabolic disorders such as dyslipidemia have been in the list of serious health problems in the modern world. This disease is considered as the major contributing factor to the mortality of men and women in both developed and developing countries. With the aims of identifying the potential plant-derived natural compounds as PPARa agonists for assissiting in drug discovery of anti-hyperlipidemia, 142 compounds were docked into the structure of PPARa (PDB ID: 5HYK). The screening results discovered 34 compounds including 20 flavonoid, 4 terpenoid and 10 alkaloid compounds strongly accommodated and formed good interactions with the residues of the PPARy binding site as the reference drugs did. Flavonoid was the best compounds attaching into the binding site of the PPARa. Top compounds for anti-dyslipidemia were identified, namely formononetin from Thermopsis alterniflora, diosmetin from Musa spp., luteolin from Elsholtzia ciliate; steviol from Stevia rebaudiana; and tuberocrooline from Stemona tuberosa. These compounds formed favorable interactions with the binding site of PPAR α through hydrogen bonds and hydrophobic interactions with key residues of the target, Ser280, Tyr314 and Tyr464, leading to agonistic activities. Combination of these results and the experimental results of these compounds for dyslipidemia effects, the current study also suggested the mechanism of the compounds as the PPAR α agonists. Therefore, this study provided useful information for further drug discovery and design of potential lipid-lowering agents as the replacement for the current PPAR α agonists like fibrate drugs.

LIST OF ABBREVIATIONS

LDL: low-density lipoprotein; HDL: high-density lipoprotein; PPARs: Peroxisome Proliferator-Activated Receptors; NPC1L1: Niemann-pick C1-like1; AF1: activation function 1; AF2: activation function 2; RMSD: root mean square deviation.

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CONFLICT OF INTEREST

The authors declare that there is no conflict of interest.

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ETHICAL CONSIDERATIONS

Not available.

AUTHORS' CONTRIBUTIONS

The authors have participated in completing the work including TLQV for docking of terpenoids; TAN for docking of saponins; PNKH for docking of alkaloids; BHGN for docking of fibrates and falvonoid compounds; and PTVN for conducting the project, supervising the process and drafting, editing and proofreading manuscript.

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